



## METHANOL

### 1. CHEMICAL IDENTITY

Chemical Name :	Methanol	Chemical Classification :	Alcohol
Synonyms :	Methyl Alcohol, wood alcohol, wood spirit, Colonial Spirit	Trade Name :	
Formula :	CH <sub>3</sub> OH	C.A.S. No. :	67-56-1
		U.N. No.:	1230

#### Regulated Identification :

Shipping Name :	Methanol		
Codes/Label :	Flammable Liquid, Class 3	Hazchem Code No. :	2 P E
Hazardous waste	17		
I.D. No. :			
Hazardous ingredients :	C. A. S. No.		
1. Methyl alcohol	67-56-1		

### 2. PHYSICAL AND CHEMICAL DATA

Boiling Range/point degreeC : 64.5	Physical State: Liquid	Appearance :Colourless, Watery
Melting/Freezing Point degree C : - 97.8		Odour : Alcoholic odour
Vapour Pressure at 35 degree C : 100	mm Hg at 21.2 degree C	
Vapour Density : 1.10 (Air = 1)	Solubility in water at 30 degree C : Miscible	Others : Miscible with Ethanol, Ether, Benzene, Ketones & Other Organic solvents. Vapours forms explosive mixtures with air and Oxygen.
Specific Gravity : 0.79 Water = 1	pH: Neutral	

### 3. FIRE AND EXPLOSION HAZARD DATA

Flammability : Yes	LEL : 6.0 %	Flash Point degree C : 16.1 (OC)	Autoignition
	UEL : 36.5 %	vapours form explosive mixture with air/oxygen.	temperature degree C
TDG Flammability : 3		Flash Point degree C : 11.11 (CC)	463.8
Explosion Sensitivity to Impact : Stable		Explosion Sensitivity to Static Electricity : Yes. Vapours are explosive	Hazardous Combustion Products : Emits acrid smoke and irritating fumes, CO.
Hazardous Polymerisation.:	Will not occur		
Combustible liquid : Yes	Explosive	Corrosive	NO
	Material : No	Material :	
Flammable Material : Yes	Oxidiser : No	Others :	No





Pyrophoric Material : No

Organic Peroxide : No

#### 4. REACTIVITY DATA

Chemical Stability :

Stable

Incompatibility with  
other material :

Strong Oxidisers, Beryllium Dihydride, Metals (K, Mg), Carbon Tetra chloride +  
Metals (Al, Mg, Zn), Oxidants.

Reactivity :

Violent reaction with alkaline aluminium salt, acetyl bromide, chloroform + sodium  
hydroxide, Nitric acid,  $\text{HClO}_4$ ,  $\text{P}_2\text{O}_3$ .

Hazardous Reaction Products :

Combustion will produce carbon monoxide and asphyxiants.

#### 5. HEALTH HAZARDS DATA

Routes of

Inhalation, Ingestion, Eyes and skin.

Entry

Effects of

Exposure/Symptoms

High concentrations can produce central nervous system depression and optic  
nerve damage. 50,000 ppm will probably cause death in 1-2 hrs. is absorbed  
through skin. Swallowing may cause death or eye damage.

Eyes Liquid may cause conjunctival irritation and transient corneal damage. vapour may  
cause conjunctival irritation.

Skin Material may cause irritation. Repeated or prolonged contact may produce defatting  
of the skin leading to irritation and dermatitis. Liquid may be absorbed through the  
skin in toxicologically significant amounts if area of contact is large and exposure  
prolonged.

Ingestion Swallowing may have the following effects : Symptoms similar to alcohol  
intoxication, Central nervous system depression, nausea, vomiting, loss of co-  
ordination, temporary or permanent blindness, coma and death.

Inhalation Exposure to vapour may have the following effects :- Headache. Exposure to vapour  
at concentrations of 1000 ppm and above may have the following effect  
Systemic effects similar to those resulting from ingestion. Because of slow  
elimination from the body repeated exposures may result in accumulation.

Emergency Treatment

Remove the victim from exposed area and apply artificial respiration if breathing  
has stopped. Induce vomiting and give 2 teaspoons of baking soda in a glass of  
water. In case of skin or eyes flush with plenty of water for 15 minutes. Seek  
medical aid.

TLV (ACGIH)

200 ppm

260  $\text{mg}/\text{m}^3$

STEL : 250 ppm, 310  $\text{mg}/\text{m}^3$

Permissible

Odour Threshold

Exposure Limit

200 ppm

260  $\text{mg}/\text{m}^3$

100 ppm, 130.87  $\text{mg}/\text{m}^3$

LD - 50 (Oral Rate)

5629  $\text{mg}/\text{kg}$

IDLH

25000 ppm, 19230  $\text{mg}/\text{m}^3$

NFPA Hazard Signals

Health

Flammability

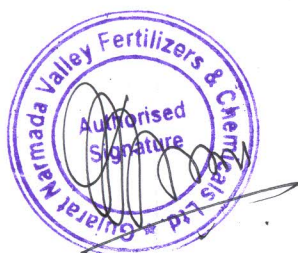
Reactivity

Special

1

3

0





## 6. PREVENTIVE MEASURES

Personal Protective Equipment	<p><b>VENTILATION :</b> Use only with adequate ventilation. Ventilate as needed to comply with exposure limit. Explosion proof ventilation Equipment required.</p> <p><b>EYE :</b> Splash proof chemical goggles or full face shield recommended to protect against splash of product.</p> <p><b>GLOVES :</b> Protective gloves recommended to protect against contact with product. The following glove materials are acceptable , Neoprene, Nitrile, Polyvinyl Alcohol, Viton.</p> <p><b>RESPIRATOR :</b> Use a positive pressure-demand full face supplied air respirator or SCBA for exposures above 50x of the exposure limit. If the exposure is above IDLH (Immediately dangerous to life and health) or there is the possibility of an uncontrolled release or exposure levels are unknown. Then use a positive pressure-demand full-face supplied air respirator with escape bottle or SCBA.</p>
Handling and Storage Precautions	<p>Use in well ventilated area. Avoid inhaling vapour. Avoid contact with eyes, skin and clothing. Keep container tightly closed when not in use.</p> <p>Storage should be cool, well ventilated. Store away from sources of heat or ignition. To avoid moisture contamination, store under a nitrogen blanket or fit a dessicant unit in a tank vent line. storage and transfer of equipment should be adequately earthed and bonded to prevent the accumulation of static charges. Storage tanks must be positioned within a bunded area. Suitable storage materials are :- mild steel, stainless steel, Do not store in aluminum and its alloys, lead zinc, certain rubbers, polystyrene. Because of its corrosive nature, extreme care should be exercised in the choice of materials for pumps, gaskets and lines. For gaskets and seals use :- compressed asbestos, butyl rubber, PTFE.</p> <p>Follow Petroleum rules 1976.</p>

## 7. EMERGENCY AND FIRST AID MEASURE

FIRE	<b>FIRE EXTINGUISHING MEDIA</b>	<p>Burns with almost invisible flames.</p> <p>Do not extinguish fires unless flow/leakage can be stopped. Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide. Keep containers and surroundings cool with water spray.</p>
FIRE	<b>Special Procedures</b>	<p>Keep the container cool by spraying water if exposed to heat or flame. Do not use water jet.</p> <p>Burns with almost invisible flame. Use "Paper on rod" detector or salt water spray to detect flame boundary if necessary.</p>
EXPOSURE	<b>Unusual Hazards First Aid Measures</b>	<p>Containers may explode in fire.</p> <p><b>EYES :</b> Immediately flood the eye with plenty of water, preferably warm, for at least 20 minutes, holding the eye open. Obtain medical attention urgently.</p> <p><b>SKIN :</b> Immediately flood the skin with large quantities of water, preferably under a shower. Remove contaminated clothing as washing proceeds. Continue washing for at least 20 minutes.</p>





#### Notes to Physician

##### Antidotes/Dosages

Never administer anything by mouth if a victim is losing consciousness, is unconscious, or is convulsing. Do not induce vomiting. Have the victim drink about atleast 240-300 ml of water to dilute stomach contents. If the vomiting occurs naturally, lean the victim forward in order to reduce risk of aspiration. Repeat administration of water. Keep warm and at rest. If there is difficulty in breathing give oxygen. If breathing stops or shows signs of failing, apply artificial respiration. Do not use mouth to mouth ventilation. If heartbeat absent, give external cardiac compression. In acute poisoning artificial respiration and alkali therapy of acidosis may be necessary as a matter of urgency. Gastric Lavage, I.V. Infusion by Sodium bicarbonate, Massive alkalization in life saving and eye saving measures. Estimate alkali reserve in blood or pH of urine and plan further treatment accordingly. Give small quantity of Ethyl alcohol every 4 hourly.

#### SPILLS

##### Steps to be taken

Contain and absorb using earth, sand or inert materials (if feasible). Transfer into suitable containers for recovery or disposal. If possible soak up remainder with absorbent material. Finally flush the area with plenty of water. Contaminated absorbent material may pose the same hazard as the spilled product. Treat contaminated water used for spill / leak control or used for dilution

##### Waste disposal Method

Incineration. Dispose of in accordance with all applicable local and national regulations. If correctly incinerated this material will decompose to carbon dioxide and water only.

**INHALATION :** Remove from exposure. Move to fresh air. Keep warm and at rest. If there is difficulty in breathing, give oxygen. If breathing stops or shows signs of failing, apply artificial respiration. If heart beat absent, give external cardiac compression. Obtain medical attention urgently..

Baking-soda in glass of water. Call on a Doctor.





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### 8. ADDITIONAL INFORMATION / REFERENCES :

A human poison by ingestion, Poison also by skin contact. The main toxic effect is extended to the nervous system, particularly optic nerves and retina, which may lead to permanent blindness. Once absorbed, it is slowly eliminated. Coma by severe exposure may last for 2-4 days. Persons with eye, liver, kidney and lung problems should avoid contact with this. Periodic medical check up is recommended. Dangerous fire hazard when exposed to heat, flame, and oxidiser.

### 9. MANUFACTURE / SUPPLIERS DATA

Name of Firm	M/S GNFC Ltd.	Contact person in emergency
Mailing Address	Po. Narmadanagar, Bharuch - 392 015	
Telephone/Fax Nos. : 47001 - 47028 / 02642-47094		
Telephonic Address	Local Bodies Involved	
	Standard	Road Tankers
	Packing	
	Tremcard	Yes
	Detail/Ref.	
	Others.	

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