CALCIUM CARBONATE

1. CHEMICAL IDENTITY

Chemical Name	Calcium Carbonate					Chemical Classification			Inorę	ganic Compound
Synonyms	Calcite					Trade Name	e		-	
Formula	CaCO₃					C.A.S. No.			471-	34-1
						U.N. No.			-	
Regulated Identifie	cation :									
Shipping Name	Calcium	Ca	rbonate							
Codes/Label	-					Hazchem C	ode No		-	
Hazardous waste I.D. No. :	16									
Hazardous ingredients	C.A.S. N	۱o.								
1. Calcium Carbonate	471-34-	1								
2. PHYSICAL			/IICAL	DATA				•		
Boiling Range/point degreeC : Melting/Freezing Point degree C :				Decom before	poses boiling	s Physical State a Solid.	tate :	: Appe		ce : white Powder
			1339 102 5 a	(at (at			Odou		ourless	
Vapour Pressure a Vapour Density : N	at 35 deg N. A.	ree So	C : Iubility i	N A n water at 3	0 degi	ee C : 0.014 K	(g/m³	Othe	rs : -	
Specific Gravity : 2	2.71	p⊢	l : 8 to 9	in water						
Water = 1										
3. FIRE AND E	EXPLOS	SIO	N HAZ		Α					
Flammability : No		No	on Flam	mable	Flash (OC)	Point degree	C : N	ot Pertir	nent	Autoignition degree C temperature :-
TDG Flammability	/ : N.A.				Flash (OC)	Point degree	• C : N	ot Pertir	nent	
Explosion Sens Impact	sitivity	to	Stable		Explo Electi	sion Sensitivity icity	y to Stat	tic : N.A	\ .	Hazardous Combustion Products : Calcium Oxide, Oxides of carbon.
Hazardous Polym	erisation		Will no	t occur.						
Combustible liquic Combustible gas.	d:No. . :No.			Explosive : Material	No (Corrosive Material	Yes	S		
Flammable Materi	ial : No.			Oxidiser : N	lo (Others				
Pyrophoric Materia	al :No			Organic Peroxide		No.				
4. REACTIVITY	Y DATA									
Chemical Stability	,			Stable un normal conditions.	der					
Incompatibility with other material Reactivity	h			Acids, Alum (Mg + H ₂) As above.	, Amm	onium Salts &	Flourin	e. React	s viole	ently with F ₂ Flourine

Hazardous Reaction Products	Calium Oxides and C	Calium Oxides and Oxides of Carbon						
5. HEALTH HAZARDS DA Routes of	ATA Inhalation, Skin, Eye	Inhalation, Skin, Eyes & Ingestion.						
Entry Effects of Exposure/Symptoms								
SKIN	Dust may cause irrit the skin.	Dust may cause irritation by abrasion, and may have a drying effect upon the skin.						
EYE	Dust may cause slig	nt transient irritation						
INGESTION	Repeated doses may	Repeated doses may cause Consipation.						
INHALATION	Exposure to dust at Irritation of Nose, thr	Exposure to dust at high concentrations may have the following effects : Irritation of Nose, throat and respiratory tract.						
SPECIAL TOXIC EFFECTS								
Emergency Treatment								
TLV (ACGIH) (Dust - respirable)	5 mg/m³ 10 mg/m³	STEL : -						
Permissible -	-	Odour Threshold						
LD	IDLH	-						
NFPA Hazard Signals He	alth Flammability	Reactivity Special						
6. PREVENTIVE MEASUR	RES							
PersonalDust MaProtectiveEYE PFEquipmentSKIN PI	ask should be worn while wo ROTECTION : Dust tight Go ROTECTION : Protective su	orking with the material. ggles. uit.						
Handling and Use loca Storage Precautions	al exhaust ventilation, If con	ditions are dusty. Store away from Acids, Water, etc.						
7. EMERGENCY AND FIF	RST AID MEASURE							
FIRE FIF ME	RE EXTINGUSTING	Not Combustile. Select extinguishing agent appropriate to other materials involved.						
FIRE Spe	ecial Procedures							
Un	usual Hazards							
EXPOSURE Fire	st Aid Measures	SKIN : Wash Skin with soap & water. EYE : Wash out eyes with plenty of water INHALATION : Bemove from exposure						
Ant	tidotes/Dosages							
Notes to Physician Tre	eat symptomatically							
SPILLS Ste	eps to be taken	Transfer into suitable container for recovery or						
Wa	aste disposal Method	This chemical occur naturaly and widely in nature. Landfill Method should be used for disposal						

8. ADDITIONAL INFORMATION / REFERENCES :

ACUTE TOXICITY : Calcium carbonate is used as an non-absorbable antacid, but high doses may be absorbed and at the same time reduce magnesium absorption. After a signle 4 g dose in normal subject there is a transient increase in plasma calcium concentration, but hypercalcaemia is more persistent after a dose of 8 g.

Dose-related constipation is commonly noted when 20-40 g calcium carbonate is taken daily, and haemouhoids may occur.

During chronic administration of calium carbonate, hypophosphataemia may result if phosphate intake is low.

IRRITANCY EYES : Single application to the rabbit eye produced minimal conjunctival irritation. 9. MANUFACTURE / SUPPLIERS DATA

Name of Firm Mailing Address		Contact Person in Emergency	M/S GNFC Ltd. P.O. Narmadanagar, Bharuch - 392 015
Telephone/Fax Nos. : 470 47094	001- 47028/02642 -		
Telephonic Address	Local Bodies Involved		
	Standard Packing		HDPE Bags
	Tremcard Detail/Ref.		Yes
	Others.		

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